

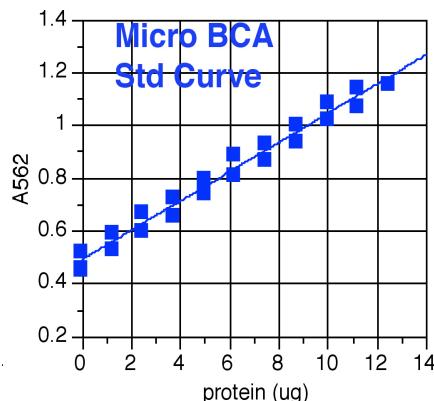
**QUIZ 3 MAY 12, 2017**  
**MCB 120L S2017V2**

Name: \_\_\_\_\_

Lab Section (circle): **MORAND A01** **LAGARIAS A02**

Acid phosphatase (APase) is an enzyme that, like alkaline phosphatase studied in Exp. 3, can catalyze p-nitrophenol phosphate hydrolysis. You have been asked to determine the specific activity of APase in an extract from the newly isolated archeabacterium *Thermococcus furiosus*. Since you have taken MCB120L, you know that you need to determine both **protein concentration** and **APase enzyme activity** of the *T. furiosus* extract. The total volume of the extract is 2.1 ml.

*T. furiosus* cells are hard to obtain in quantity since it only grows near volcanic vents deep in the ocean. For this reason, you used a Micro BCA Protein Assay in which BSA standard and Babelfish extracts were diluted into 50  $\mu$ l (final volume) from which 12.5  $\mu$ l was removed and diluted with 237.5  $\mu$ l BCA reagent in individual wells of a microplate. This colorimetric assay measures the change in absorbance at 562 nm. You analyzed different amounts of your undiluted *T. furiosus* extract and obtained the results shown in Table 1 below. The BCA standard curve that uses BSA as the protein standard is shown below on the right. The formula for the best-fit line of this data is  $y = 0.0554x + 0.488$  where y corresponds to  $A_{562}$  and x corresponds to  $\mu$ g protein.



**Table 1. BCA Protein Assay**

Sample Mixtures		Assay Results	
APase extract assayed, $\mu$ l	Amount of water added, $\mu$ l	Ave $A_{562}$ for 12.5 $\mu$ l sample mix + 237.5 $\mu$ l BCA reagent (n=3)	$\mu$ g protein in assay
2	48	0.500	
4	46	0.712	
8	42	0.942	
16	34	1.282	

**1. (2 pts) Complete the last column of data table above.** In the box below, indicate the **concentration** of APase protein in your *T. furiosus* extract (mg/ml) and the **total protein** (mg) in the *T. furiosus* extract. Show your work below or on back.

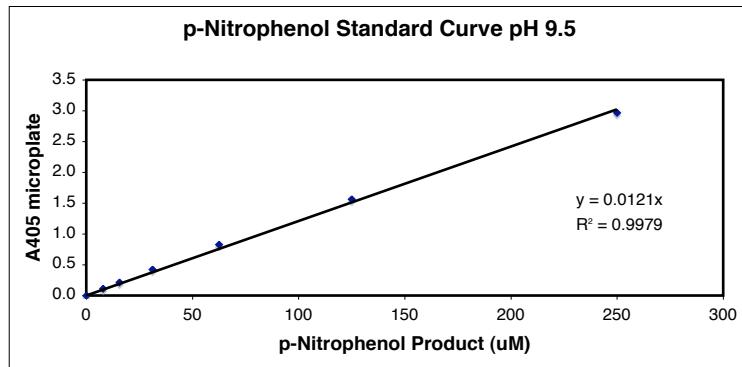
Conc =

Total Protein =

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You next assayed the *T. furiosus* extract for APase activity at four pHs using the **continuous** and microplate **fixed time** assays protocol **identical** to those you used in Experiment 3A&B, **except that the fixed time was shortened to 4 min**. The assay results for **1/50 dilutions** of the *T. furiosus* extract are shown below. Since acidic buffers were used for these assays, the pH of the fixed time assays were measured after the 1.0 M K<sub>2</sub>HPO<sub>4</sub> Stop solution was added to permit correction for the ionization of p-nitrophenol product in the microplate. This was done with a micro pH electrode and the pHs measured are shown in the Table. The standard curve for p-nitrophenol for the microplate is shown below.

Continuous Assay (1 cm cuvette)					Microplate Fixed Time Assay					
Assay pH	Non-enz. control $\Delta A_{410} \text{ min}^{-1}$	+ Enz. $\Delta A_{410} \text{ min}^{-1}$	Corr. $\Delta A_{410} \text{ min}^{-1}$	I.U. per assay	Assay pH	Measured pH after 1.0 M K <sub>2</sub> HPO <sub>4</sub> Stop solution added	Non-enz. $\Delta A_{410} (4 \text{ min})^{-1}$	+ Enz. $\Delta A_{410} (4 \text{ min})^{-1}$	Corr. $\Delta A_{410} (4 \text{ min})^{-1}$	I.U. per assay
4	0.0003	0.0005	0.0002	XXX	4	7.0	0.010	0.4765		
5	0.0002	0.0020	0.0018	XXX	5	7.5	0.050	0.6980		
6	0.0001	0.0100	0.0099	XXX	6	8.0	0.080	0.5728		
7	0.0002	0.0753	0.0751		7	8.5	0.102	0.4609		



**2. (3 pts)** Based on the continuous assay results, calculate the specific activity of the *T. furiosus* APase at **pH 7**. Put IU per assay in Table above. Show how you calculated this value below and place your calculated specific activity in the box below.

APase sp. activity @ pH 7 =

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**3. (4 pts) Complete the Table on the previous page.** In the space below (and continued on next page or back), show how you calculated the I.U. per assay for the **fixed time** assays measured at pH 7. Put answers in Table on page 1. All other calculations can be performed on back pages.

**4. (1 pt)** Based on your data (above), at which pH is the activity of APase the largest – **pH 4, pH 5, pH 6 or pH 7** (CIRCLE ANSWER)? Justify your answer below.